

Kinetic studies of the OH(X²Π) and O(³P) initiated reactions with selected short chain iodoalkanes

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A Flash Photolysis–Resonance Fluorescence (FP-RF) technique was used to investigate the kinetics of the OH(X²Π) radical initiated reactions with selected iodoalkanes (RI), namely CH₃I, CH₂I₂, C₂H₅I, n-C₃H₇I, iso-C₃H₇I, CHI₃ and O(³P) radical initiated reactions with selected RI, namely CHI₃ and C₂H₅I. The reactions of OH(X²Π) radicals with RI were studied over the temperature range 295 – 390 K and pressure around 200 Torr of He.¹⁻³ The reaction of the OH(X²Π) radical with CHI₃ was studied at T=298K only. The reactions of O(³P) radical with CHI₃ and C₂H₅I were studied over the temperature range 296 – 373K in 14 Torr of He. The experiments involved time-resolved RF detection of OH (A²Σ⁺→X²Π transition at λ=308 nm) and of O(³P) (λ=130.2, 130.5, and 130.6 nm) following FP of the H₂O/He, H₂O/RI/He, O₃/He and O₃/RI/He mixtures. The OH(X²Π) and O(³P) radicals were produced by FP in the vacuum-UV at wavelengths λ>120 nm using a Xe flash lamp. Measured rate coefficients for the reactions of OH(X²Π) and O(³P) radicals with RI are described by the following Arrhenius expressions (units are cm³molecule⁻¹s⁻¹):

$$k_{\text{OH}+\text{CH}_3\text{I}} = (4.1 \pm 2.2) \times 10^{-12} \exp[(-1240 \pm 200)\text{K}/T]$$

$$k_{\text{OH}+\text{CH}_2\text{I}_2} = (4.2 \pm 0.5) \times 10^{-11} \exp[(-670 \pm 20)\text{K}/T]$$

$$k_{\text{OH}+\text{CHI}_3} = (1.6 \pm 0.1) \times 10^{-11}$$

$$k_{\text{OH}+\text{C}_2\text{H}_5\text{I}} = (5.6 \pm 3.2) \times 10^{-12} \exp[(-830 \pm 90)\text{K}/T]$$

$$k_{\text{OH}+\text{n-C}_3\text{H}_7\text{I}} = (1.7 \pm 0.9) \times 10^{-11} \exp[(-780 \pm 90)\text{K}/T]$$

$$k_{\text{OH}+\text{iso-C}_3\text{H}_7\text{I}} = (7.6 \pm 3.7) \times 10^{-12} \exp[(-530 \pm 80)\text{K}/T]$$

$$k_{\text{O}+\text{CHI}_3} = (1.8 \pm 2.8) \times 10^{-12} \exp[(+430 \pm 260)\text{K}/T]$$

$$k_{\text{O}+\text{C}_2\text{H}_5\text{I}} = (2.0 \pm 1.4) \times 10^{-11} \exp[(+140 \pm 110)\text{K}/T]$$

The implications of the reported kinetic results for understanding the degradation mechanisms of iodoalkanes in case of a nuclear power plant accident are discussed. Further, the OH radical and O atom attacks on RI, namely OH-addition, H-atom and I-atom abstraction reaction channels, are discussed.

References

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