

Nighttime chemistry: Field measurements N<sub>2</sub>O<sub>5</sub> uptake and ClNO<sub>2</sub> production

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Nighttime chemistry accounts for up to half of the NO<sub>x</sub> removal from the atmosphere and produces atomic chlorine in the early morning through photolysis of ClNO<sub>2</sub>. NO<sub>x</sub> is removed through the production and heterogeneous loss of N<sub>2</sub>O<sub>5</sub>. The heterogeneous loss of N<sub>2</sub>O<sub>5</sub> typically produces nitric acid which is removed through deposition. However, it has recently been discovered that in the presence of aerosol chloride, the N<sub>2</sub>O<sub>5</sub> uptake produces ClNO<sub>2</sub>. The rate N<sub>2</sub>O<sub>5</sub> heterogeneous loss has been the subject of several laboratory studies; however, it has rarely been studied using ambient measurements. In this work, N<sub>2</sub>O<sub>5</sub> uptake coefficients are determined from ambient wintertime measurements at the BAO tower in Erie, CO. The uptake coefficients are determined using an iterative box model. These uptake coefficients were found to be anti-correlated with the nitrate fraction of aerosol confirming suppression of uptake by aerosol nitrate. Additionally, a plume of chloride was observed in measurements of aerosol chloride, HCl and ClNO<sub>2</sub>. The N<sub>2</sub>O<sub>5</sub> uptake coefficient was enhanced in this plume due to the competition between aerosol chloride and nitrate to react with N<sub>2</sub>O<sub>5</sub>.